

## Chapter 20 Study Guide Redox Reactions Answers

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### Chapter 20 Study Guide Redox

634 Chapter 20 Redox Reactions CHAPTER 20 What You'll Learn You will examine the processes of oxidation and reduction in electron-transfer reactions. You will discover how oxidation numbers of elements in compounds are determined and how they relate to electron transfer. You will separate redox reactions into their oxidation and reduction processes.

### Chapter 20: Redox Reactions - Neshaminy School District

Chapter 20 & 21 Vocabulary (Redox) oxidation. reduction. oxidation-reduction reaction (redox) reducing agent. complete or partial loss of electrons between reactants. complete or partial gain of electrons or loss of oxygen. reactions that involve the process of oxidation and reduction. the substance that loses electrons.

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### 20.2 Balancing Redox Reactions - Chapter 1 | Dr. Fus

Chapter 20 Worksheet: Redox ANSWERS I. Determine what is oxidized and what is reduced in each reaction. Identify the oxidizing agent and the reducing agent, also. 1.  $2\text{Sr} + \text{O}_2 \rightarrow 2\text{SrO}$  Sr 0 to  $\text{Sr}^{2+}$ ; oxidized/reducing agent  $\text{O}_0$  to  $\text{O}^{2-}$ ; reduced/ox. ag. 2.  $2\text{Li} + \text{S} \rightarrow \text{Li}_2\text{S}$  Li 0 to  $\text{Li}^{+}$ ; oxidized/red. ag.  $\text{S}_0$  to  $\text{S}^{2-}$ ; reduced/ox. ag. 3.

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Each of the following regarding redox couples is true EXCEPT: a. standard reduction potentials are measured relative to a standard hydrogen cell ( $\text{H}^+/\text{H}_2$ ) b. for cells in which electrons flow toward the sample half-cell, the reduction potential is positive c. a redox couple consists of a substance being oxidized, and the substance oxidizing it

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120 Chemistry: Matter and Change • Chapter 20 Block Scheduling Lesson Plans Half-Reactions pages 650–653 BLOCK SCHEDULE LESSON PLAN 20.3 Objectives • Recognize the interdependence of oxidation and reduction processes. • Derive oxidation and reduction half-reactions from a redox equation. • Balance redox equations by the half-reaction method.

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Redox in Electrochemistry Electrochemistry is the study of the redox processes by which chemical energy is converted to electrical energy and vice versa. Electrochemical processes are useful in industry and critically important for biological functioning.

### Chapter 20: Electrochemistry

The Redox Reactions chapter of this Glencoe Chemistry - Matter and Change textbook companion course helps students learn the essential chemistry lessons of redox reactions.

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