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Holt ChemFile: Problem-Solving Workbook 259pH. pH. In 1909, Danish biochemist S. P. L Sørensen introduced a system in which acidity was expressed as the negative logarithm of the H concentration. In this way, the acidity of a solution having H concentration of 10⁻⁵M would have a value of 5.

Skills Worksheet Problem Solving

Holt ChemFile: Problem-Solving Workbook 48 Mole Concept Mole Concept Suppose you want to carry out a reaction that requires combining one atom of ... Yes; the answer has the correct units of atoms of boron. Is the number of significant figures correct? Yes; the mass of boron was given to three significant figures. 1 mol B

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Holt ChemFile: Problem-Solving Workbook 112 Limiting Reactants Limiting Reactants At the beginning of Chapter 8, a comparison was made between solving stoi- ... Is the answer reasonable? Yes; you can see that 6.3 mol HCl is more than is needed to react with 2.8 mol Ca(OH)₂. Practice 1.

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Holt ChemFile: Problem-Solving Workbook 9 Conversions Name Class Date Problem Solving continued Rewrite the given quantity in simple units as follows. m³ m m m COMPUTE 0.50 m³ neon 0.50 (m m m) 10 1 0 m cm 10 1 0 m cm 10 1 0 m cm 500 000 cm³ neon EVALUATE Are the units correct? Yes; cubic centimeters were the units sought. Is the answer reasonable?

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Holt ChemFile: Problem-Solving Workbook 57 Mole Concept Name Class Date Problem Solving continued Is the answer reasonable? Yes; 2 g of boron is about 1/5 of the molar mass of boron. Therefore, 2.00 g boron will contain about 1/5 of an Avogadro's constant of atoms.

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